

# Nano tech Hmw

Chap 2  
(1, 3, 7, 10)

2.1: energy of photon:  $E = h\nu = \frac{hc}{\lambda} = \frac{1240 \text{ eV}\cdot\text{nm}}{650 \text{ nm}} = 1.91 \text{ eV}$   
 or  $3.06 \times 10^{-19} \text{ J}$   
 energy of electron: (non-relativistic)

$$E_k = \frac{p^2}{2m_e} = \frac{h^2}{2m_e\lambda^2} = \frac{(hc)^2}{2m_e c^2 \cdot \lambda^2} = \left(\frac{1240 \text{ eV}\cdot\text{nm}}{650 \text{ nm}}\right)^2 \cdot \frac{1}{2 \times 0.511 \text{ MeV}}$$

$$= 3.56 \times 10^{-6} \text{ eV} \quad \text{or} \quad 5.705 \times 10^{-25} \text{ J}$$

( relativistic:  $E_k = \sqrt{p^2 c^2 + m_e^2 c^4} - m_e c^2$   ~~$\ll m_e c^2$~~   
 $p \cdot c = \frac{hc}{\lambda} = E_{\text{photon}} = 1.91 \text{ eV} \ll m_e c^2 = 0.511 \text{ MeV}$   
 $\rightarrow$  yield same answer as non-relativistic )

2.3:  $E = h\nu = 6.63 \times 10^{-34} \times \begin{cases} 60 \text{ Hz} & = 3.978 \times 10^{-32} \text{ J} = 2.481 \times 10^{-13} \text{ eV} \\ 2.4 \times 10^9 \text{ Hz} & = 1.59 \times 10^{-24} \text{ J} = 9.926 \times 10^{-6} \text{ eV} \\ 3 \times 10^{16} \text{ Hz} & = 1.988 \times 10^{-17} \text{ J} = 124.07 \text{ eV} \end{cases}$

2.7: wave length of baseball:  
 $\lambda = \frac{h}{p} = \frac{h}{mv} = \frac{6.626 \times 10^{-34}}{0.15 \text{ kg} \cdot 90 \text{ m/s}} = \text{~~1.767~~ } 1.767 \times 10^{-31} \text{ (m)}$   
 The scale of baseball bats is much larger than wave length  
 $\rightarrow$  diffraction not happen.

2.10: electron:  $\lambda = \frac{h}{\sqrt{2m_e E}} = \frac{1240.7}{\sqrt{2 \times 4 \times 0.511 \times 10^6}} \text{ (nm)} = 6.14 \times 10^{-10} \text{ m}$   
 $E = 4 \text{ eV}$  proton:  $\lambda = \frac{h}{\sqrt{2m_p E}} = \frac{1240.7}{\sqrt{2 \times 1.67 \times 10^{-27} \times 4}} \text{ m} = 1.431 \times 10^{-11} \text{ m}$   
 photon:  $\lambda = \frac{hc}{E} = \frac{1240.7 \text{ eV}\cdot\text{nm}}{4 \text{ eV}} = 310.175 \text{ nm} = 3.10175 \times 10^{-7} \text{ m}$

all three particles have same frequency  $f = \frac{E}{h} = \text{~~9.677~~ } 9.677 \times 10^{14} \text{ (Hz)}$

momentum: electron:  $p_e = \sqrt{2m_e E} = 1.08 \times 10^{-24} \text{ (kg}\cdot\text{m}\cdot\text{s}^{-1})$   
 proton:  $p_{\text{proton}} = \sqrt{2m_p E} = 4.63 \times 10^{-23} \text{ (kg}\cdot\text{m}\cdot\text{s}^{-1})$   
 photon:  $p_{\text{photon}} = \frac{E}{c} = 2.136 \times 10^{-27} \text{ (kg}\cdot\text{m}\cdot\text{s}^{-1})$

Chap 4  
(4, 5, 6)

$$4.4: E = \frac{\hbar^2 \pi^2}{2m} \left( \frac{n_x^2}{L_x^2} + \frac{n_y^2}{L_y^2} + \frac{n_z^2}{L_z^2} \right) = \frac{\hbar^2 \pi^2}{2mL^2} (n_x^2 + n_y^2 + n_z^2)$$

$$E = 5 \text{ eV} \rightarrow n^2 = n_x^2 + n_y^2 + n_z^2 = \frac{5 \text{ eV}}{\frac{\hbar^2 \pi^2}{2mL^2}} = \frac{5 \text{ eV} \cdot 2mL^2}{\hbar^2 \pi^2} = 1.33 \times 10^{21}$$

$$E_{2,1,1} - E_{1,1,1} = \frac{\hbar^2 \pi^2}{2mL^2} (2^2 - 1^2) = \frac{3\hbar^2 \pi^2}{2mL^2} = 1.807 \times 10^{-39} \text{ J} = 1.128 \times 10^{-20} \text{ eV}$$

4.5 L changed to  $10^{-9} \text{ m}$ ,

$$(1) n^2 = \frac{E \cdot 2mL^2}{\hbar^2 \pi^2} = 13.3$$

$$(2) E_{2,1,1} - E_{1,1,1} = \frac{3\hbar^2 \pi^2}{2mL^2} = 1.807 \times 10^{-19} \text{ J} = 1.128 \text{ eV}$$

4.6 "observe clear energy discretization"  $\rightarrow \frac{\hbar^2 \pi^2}{2mL^2}$  is comparable to E  
 $\frac{\hbar^2 \pi^2}{2mL^2} \sim E = 2.5 \text{ eV} \Rightarrow L \sim \sqrt{\frac{\hbar^2 \pi^2}{2mE}} = 0.3878 \times 10^{-9} \text{ m} = 0.3878 \text{ nm}$

Alternatively:

$$L \sim \lambda = \frac{h}{\sqrt{2mE}} = 0.776 \text{ nm}$$

For proton:

$$L_p \sim \sqrt{\frac{\hbar^2 \pi^2}{2m_p E}} = 9.05 \times 10^{-12} \text{ m} = 9.05 \times 10^{-3} \text{ nm}$$

$$\text{or } L_p \sim \lambda_p = \frac{h}{\sqrt{2m_p E}} = 1.81 \times 10^{-11} \text{ m} = 1.81 \times 10^{-2} \text{ nm}$$

Chap 7 7.2  $k_B T \ll \frac{e^2}{2C}$

$$C = 0.5 \times 10^{-16} \text{ F} \quad T \ll \frac{e^2}{k_B \cdot 2C} = 1859 \text{ K}$$

$$C = 1.2 \times 10^{-12} \text{ F} \quad T \ll \frac{e^2}{k_B \cdot 2C} = 7.746 \times 10^{-4} \text{ K}$$

$$7.14: (a) \lambda = \frac{h}{\sqrt{2m_e E}} = \frac{1240}{\sqrt{2 \times 5 \times 0.511 \times 10^6}} = \frac{1240}{\sqrt{511}} \times \frac{1}{100} = \frac{12.4}{\sqrt{511}} = 5.485 \times 10^{-10} \text{ m}$$

$$(b) \text{ energy difference between two levels } = 0.5485 \text{ nm}$$

should not be smaller than E

$$\rightarrow \frac{\hbar^2 \pi^2}{2mL^2} \sim E, \quad L \sim \frac{\lambda}{2} = 0.274 \text{ nm}$$

(students who give answer like  $L \sim \lambda$  or  $L \leq \lambda$  are also correct)

(c) ~~Yes, to observe clear Coulomb blockade, number in quantum dot, energy state~~

7.14 (c): ~~to have discrete number of electrons on the island~~  
 energy gap between two levels must be of the size  $\frac{e^2}{2C}$  (charging  $E_C$ )  
 (and much larger than thermal energy  $k_B T$ ), ~~then~~ then  
 "Coulomb staircase" could be observed, ~~otherwise, for~~

Otherwise, Coulomb blockade can be observed as long as ~~the~~  
 certain conditions for  $T, R_C$  are satisfied.

So no, it's not necessary to require energy level ~~to be~~ quantized  
 for normal Coulomb blockade, but it's ~~not~~ necessary for  
 observing "Coulomb staircase" in quantum dot circuit.

Not Necess. Coulomb gap is there even if  $L \rightarrow \infty$

8.1:

~~to~~  $n$   $\Rightarrow \Delta E \rightarrow 0$

$$N_{E \leq E_n} = \text{Vol. of sphere of } n = \frac{4\pi n^3}{3} \quad \text{as } n_{x,y,z} = 0, \pm 1, \pm 2, \dots$$

$$E_n = \frac{2\hbar^2 \pi^2}{m_e^* L^2} n^2 \quad \text{so no } \frac{1}{8} \text{ factor here.}$$

$$\rightarrow N_{E \leq E_n} = \frac{4\pi}{3} \cdot \left( \frac{E \cdot m_e^* L^2}{2\hbar^2 \pi^2} \right)^{\frac{3}{2}}$$

$$N(E) = \frac{dN_{E \leq E_n}(E)}{dE} = \frac{4\pi}{3} \cdot \frac{3}{2} \cdot \left( \frac{m_e^* L^2}{2\hbar^2 \pi^2} \right)^{\frac{3}{2}} E^{\frac{1}{2}}$$

$$= 2\pi \cdot \frac{m_e^{*\frac{3}{2}} L^3}{2^{\frac{3}{2}} \hbar^3 \pi^3} E^{\frac{1}{2}}$$

consider spin degeneracy:

$$N'(E) = 2 * N(E) \text{ above}$$

$$= 4\pi \cdot \frac{m_e^{*\frac{3}{2}} L^3}{2^{\frac{3}{2}} \hbar^3 \pi^3} E^{\frac{1}{2}} = \frac{2^{\frac{1}{2}} m_e^{*\frac{3}{2}} L^3}{\pi^2 \hbar^3} E^{\frac{1}{2}}$$

set  $L=1$  and replace  $E$  with  $E - V_0$

we have same formula as (8.6):

$$N(E) = \frac{2^{\frac{1}{2}} \cdot m_e^{*\frac{3}{2}}}{\pi^2 \hbar^3} (E - V_0)^{\frac{1}{2}}$$